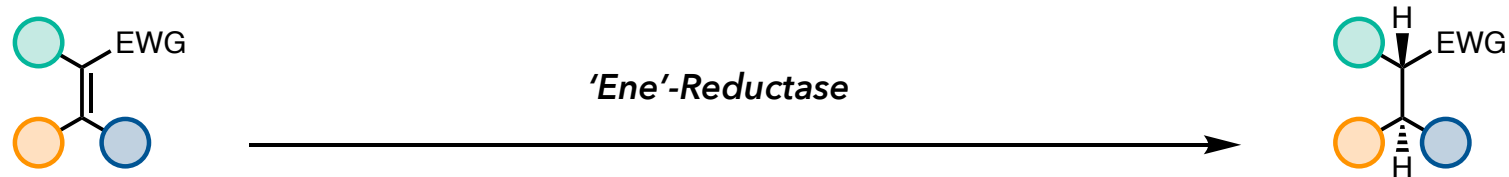
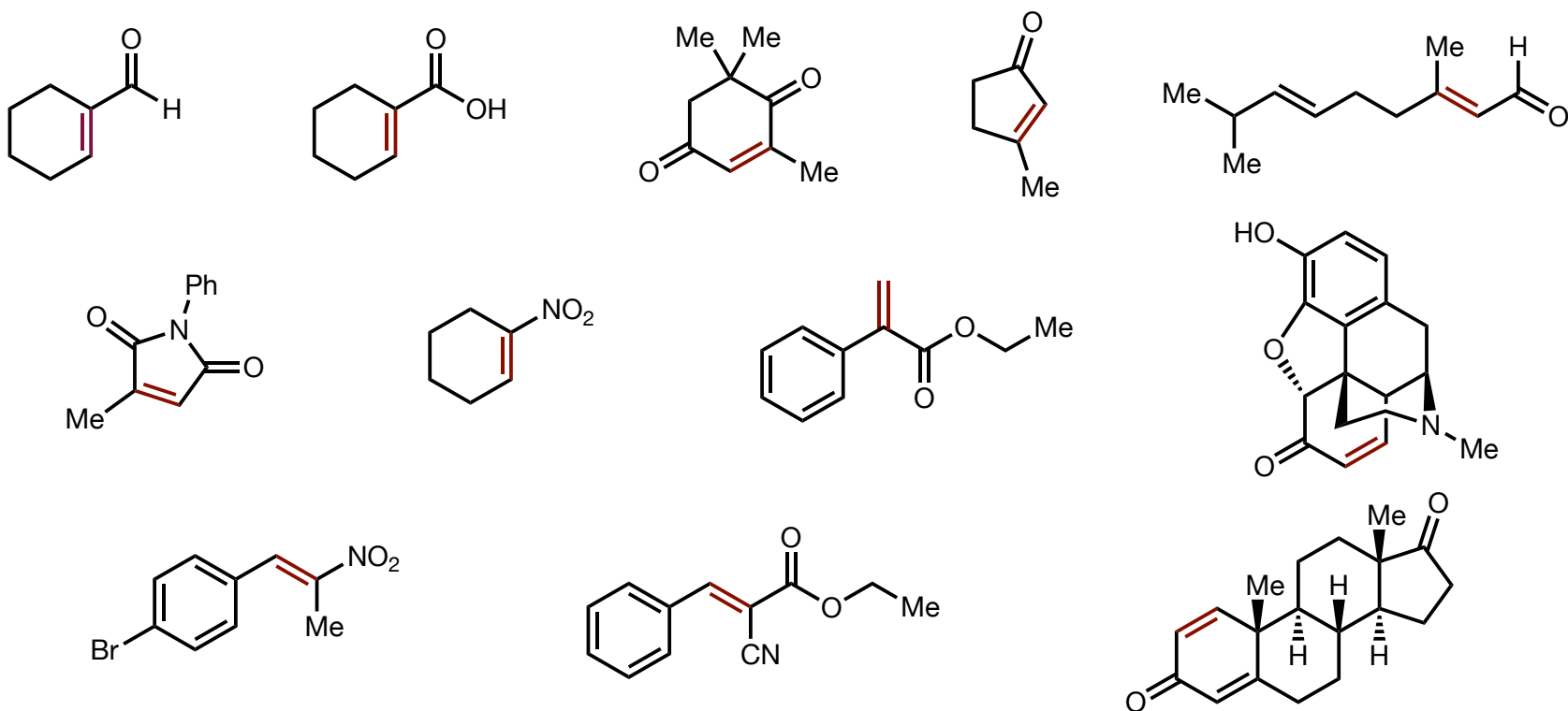


Ene-Reductases for Hydride Delivery

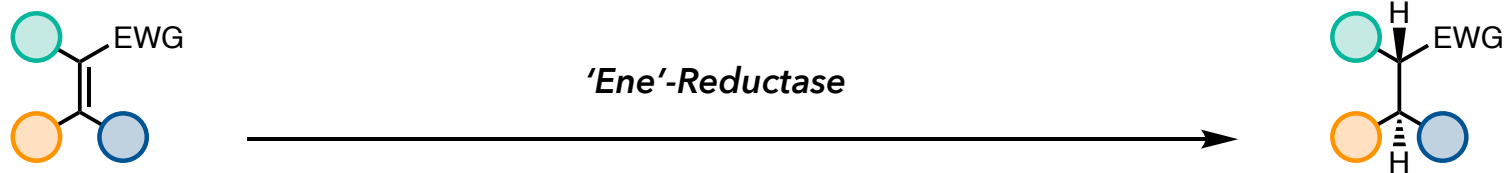


Ene-Reductases (EREDs) catalyze the asymmetric reduction of α,β -unsaturated carbonyl compounds

■ One of the most broadly substrate permissive enzyme classes across the family



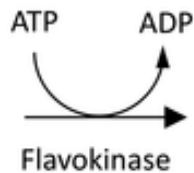
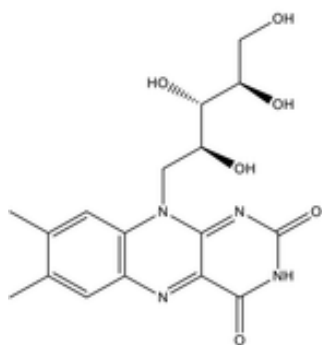
Ene-Reductases for Hydride Delivery



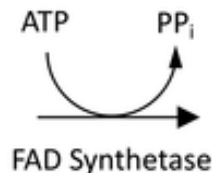
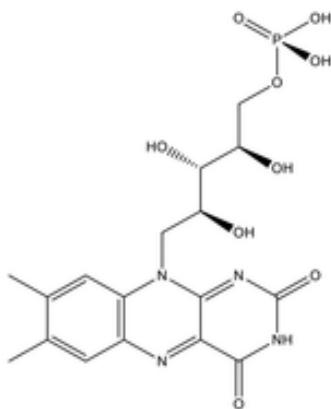
Ene-Reductases (EREDs) catalyze the asymmetric reduction of α,β -unsaturated carbonyl compounds

■ Reliant on one of three flavin analogues for reactivity

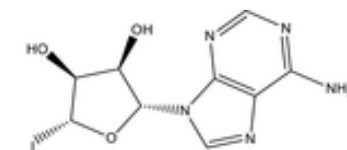
Riboflavin:
also known as
vitamin B₂



FMN: essential for enzymes such as
pyridoxamine-5-phosphate oxidase
and dihydroorotate dehydrogenase



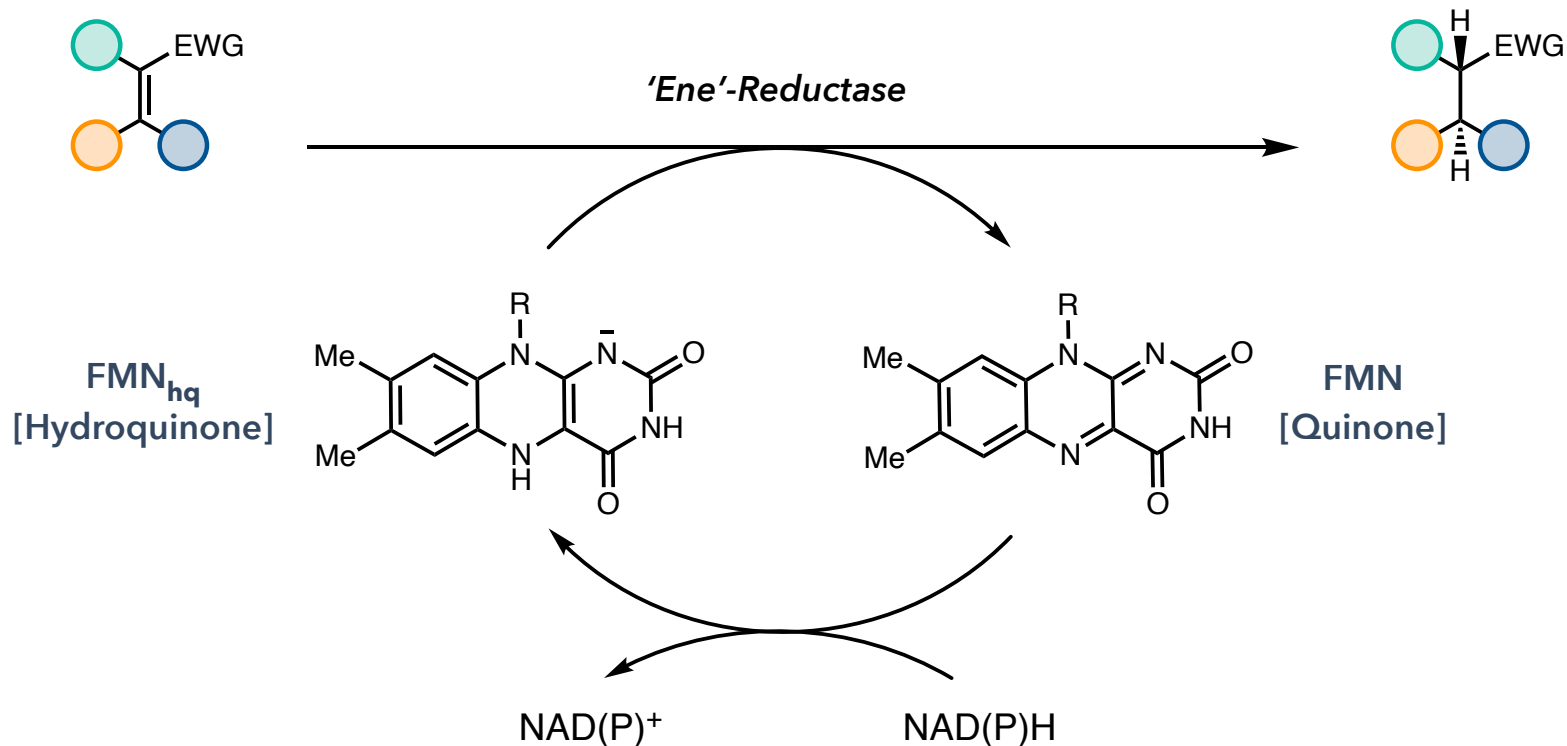
FMN = Flavin Mononucleotide



FAD: essential for enzymes
such as glutathione reductase
and thioredoxin reductase

FAD = Flavin Adenine Dinucleotide

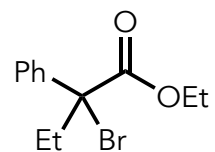
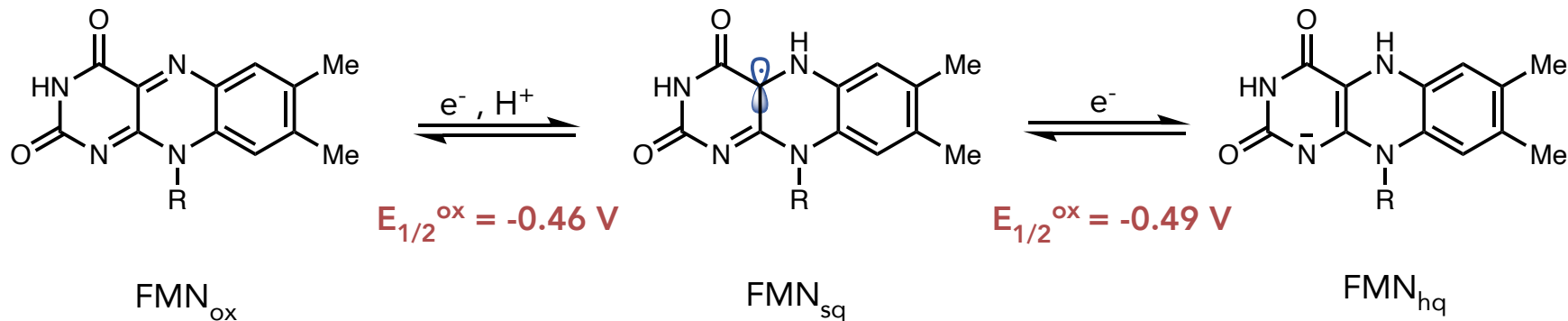
Ene-Reductases for Hydride Delivery



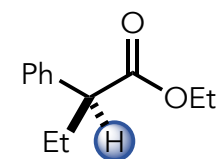
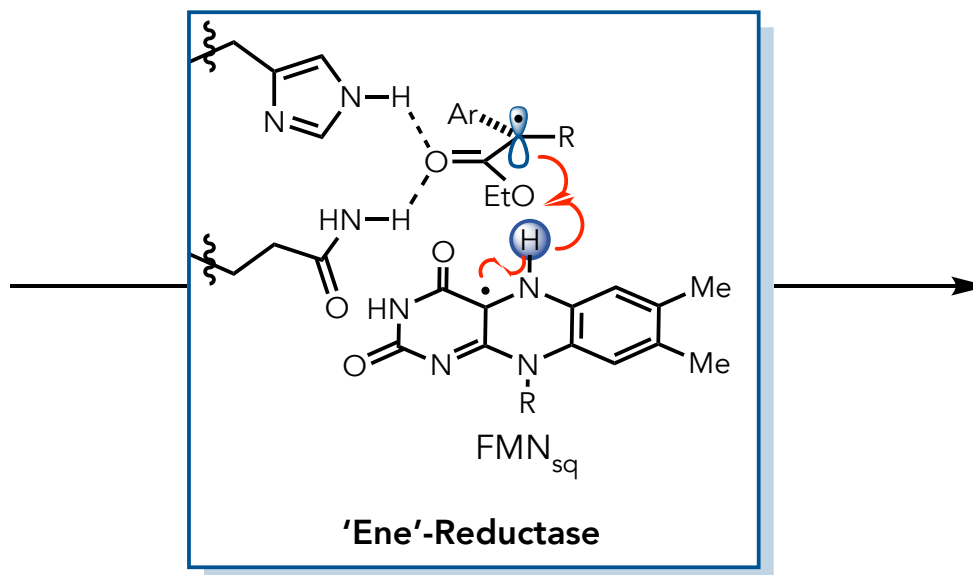
- EREDs primarily toggle between hydroquinone and quinone states in the ERED mechanism
- Flavin is reduced by the NAD(P)/NAD(P)H redox couple
 - For most cases, NAD(P)/NAD(P)H binds to the ERED and performs reduction of flavin
 - Flavin turnover systems based on single electron transfer are also very common (e-chem, photoredox)

Flavin: A Mechanistically Versatile Catalyst

FMN Redox Couples



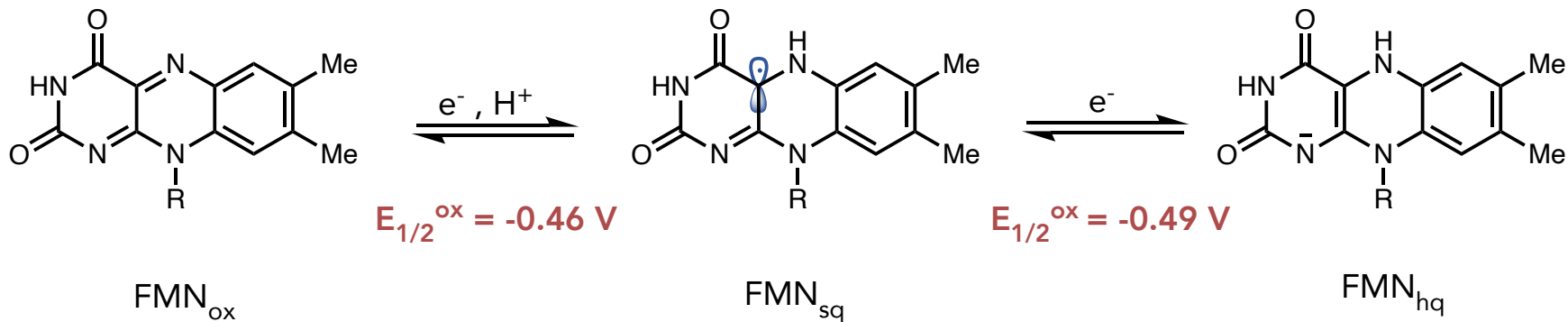
racemic



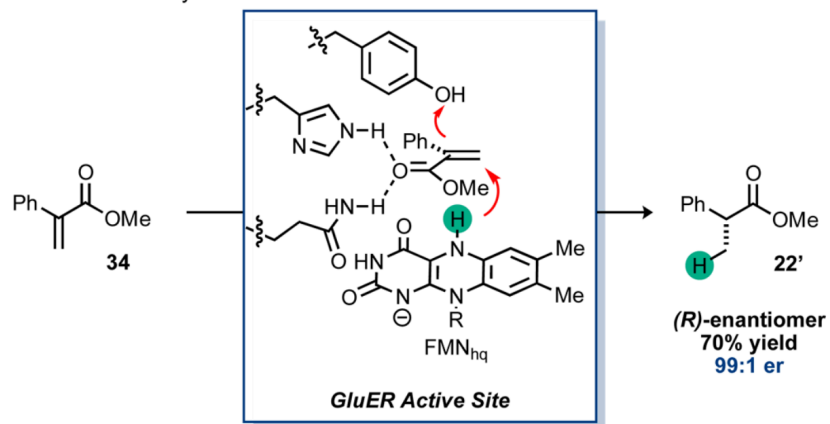
75% yield
98:2 er
16 examples
up to 98:2 er

Flavin: A Mechanistically Versatile Catalyst

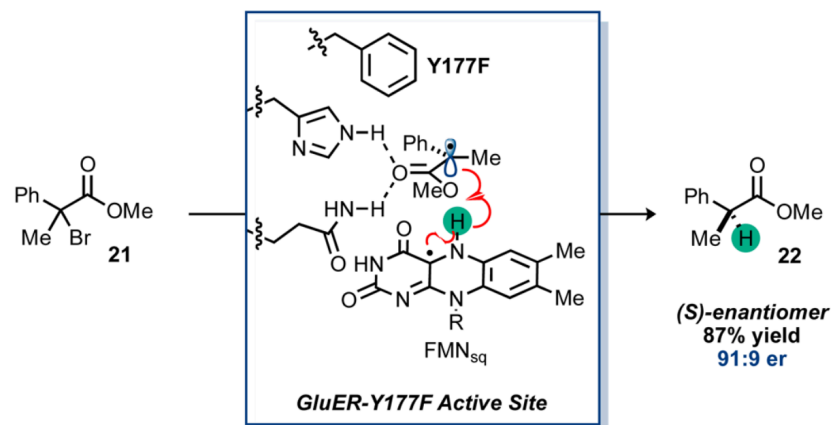
FMN Redox Couples



a. Native Reactivity

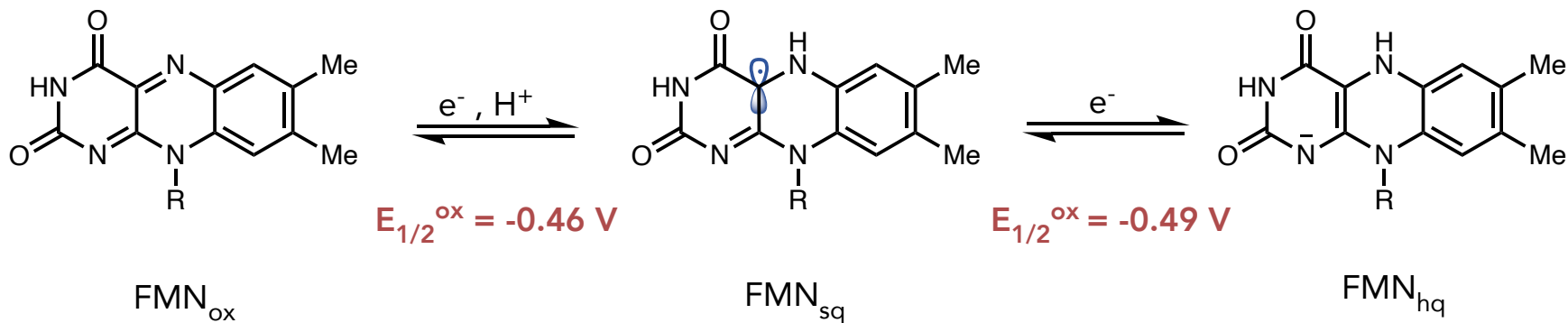


b. Non-Natural Reactivity



Flavin: A Mechanistically Versatile Catalyst

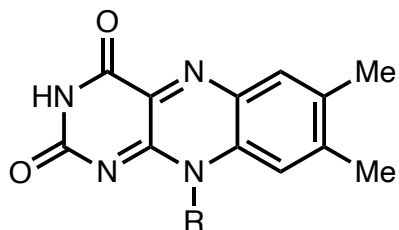
FMN Redox Couples



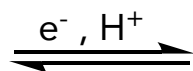
How do we expand the redox window of flavin to access substrates that are more challenging to reduce?

Flavin: A Mechanistically Versatile Catalyst

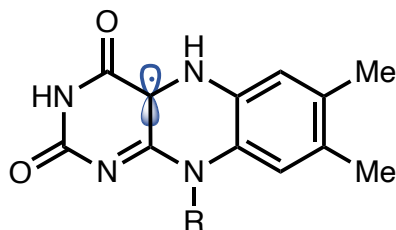
Expanded Redox Versatility



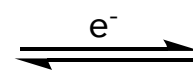
FMN_{ox}



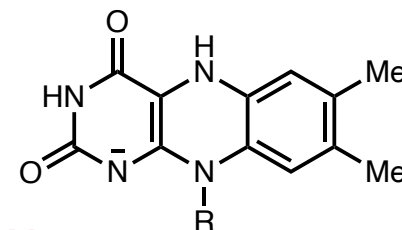
$$E_{1/2}^{\text{ox}} = -0.46 \text{ V}$$



FMN_{sq}



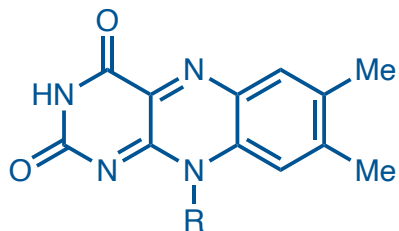
$$E_{1/2}^{\text{ox}} = -0.49 \text{ V}$$



FMN_{hq}



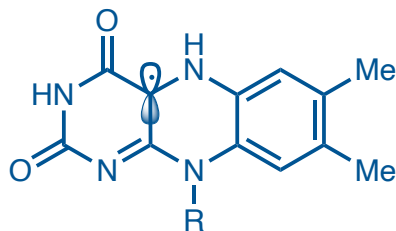
visible light



FMN_{ox}^{*}



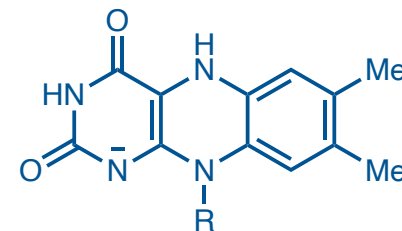
visible light



FMN_{sq}^{*}

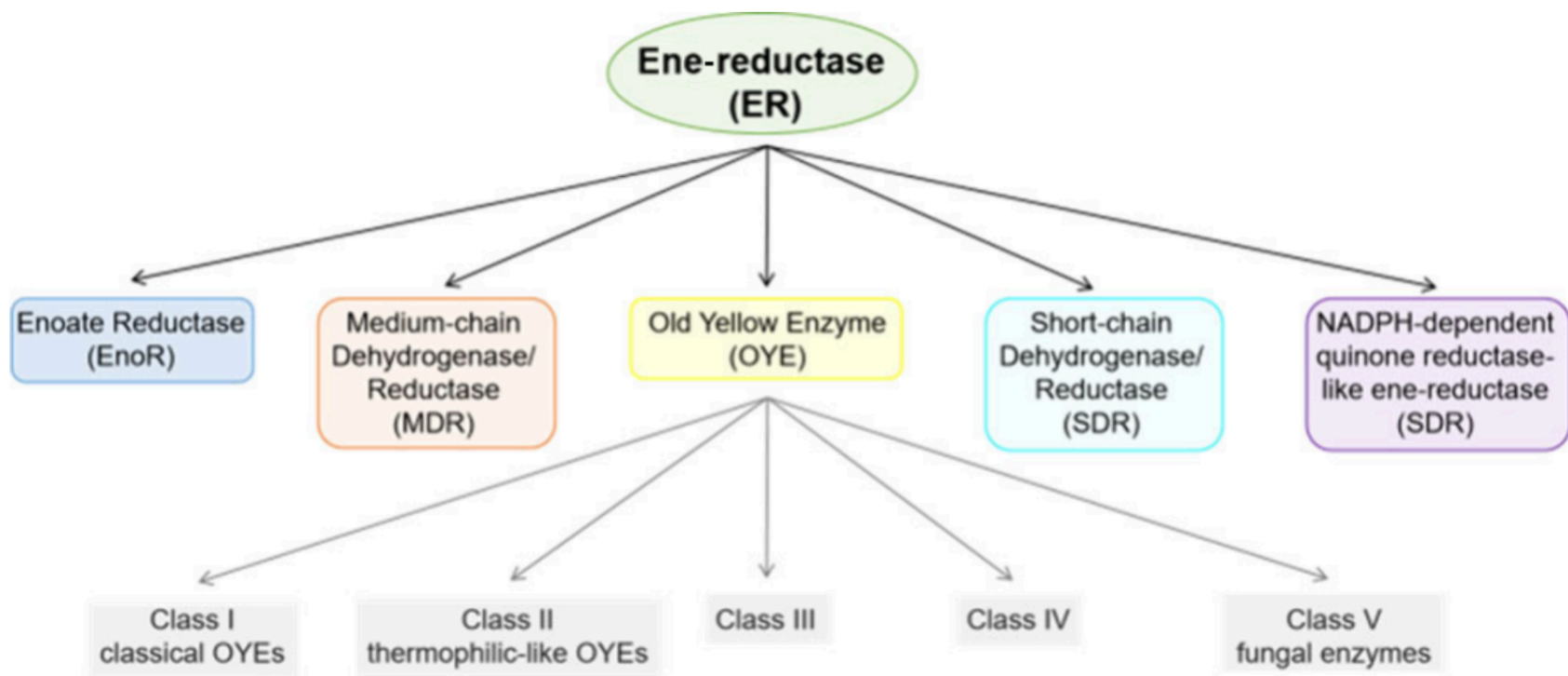


visible light

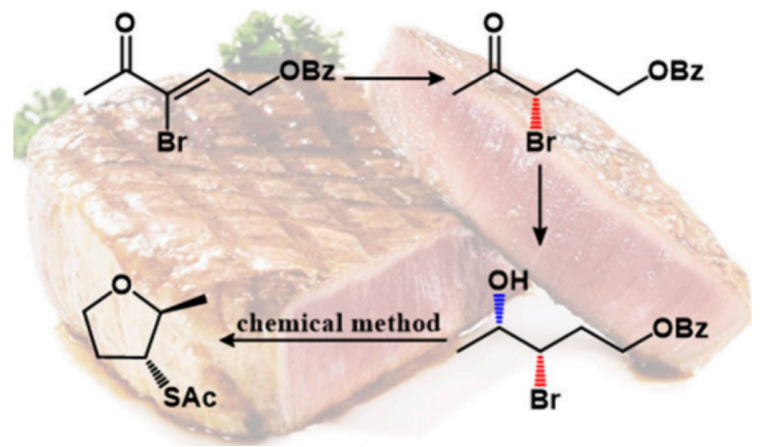
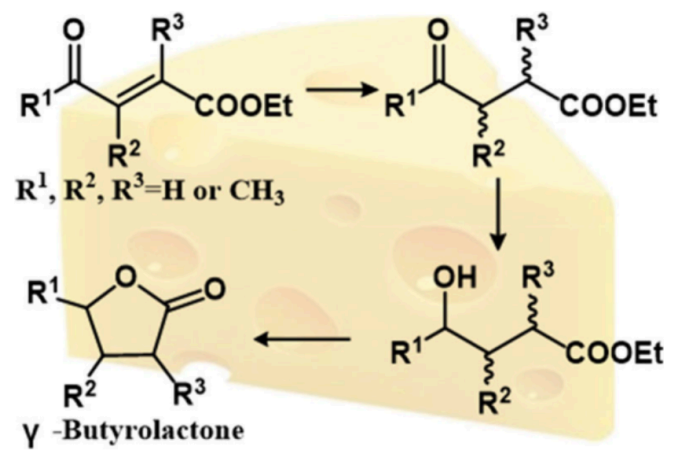
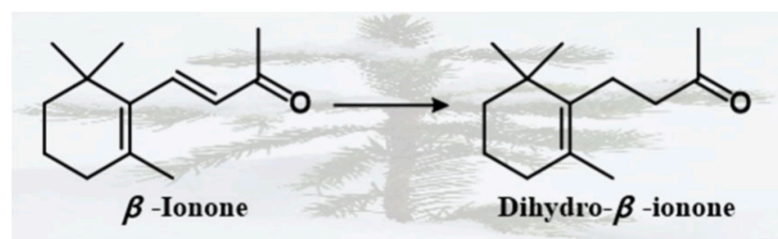
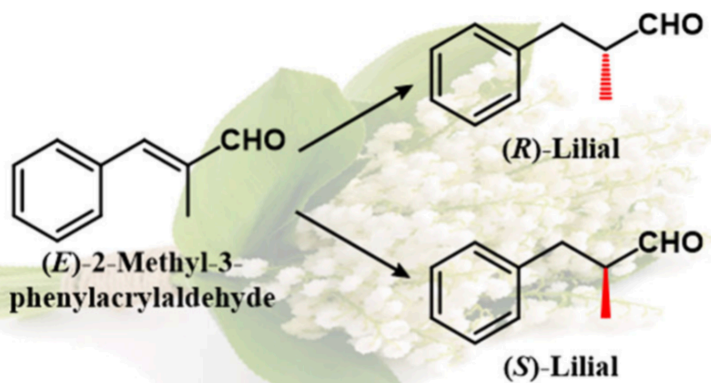


FMN_{hq}^{*}

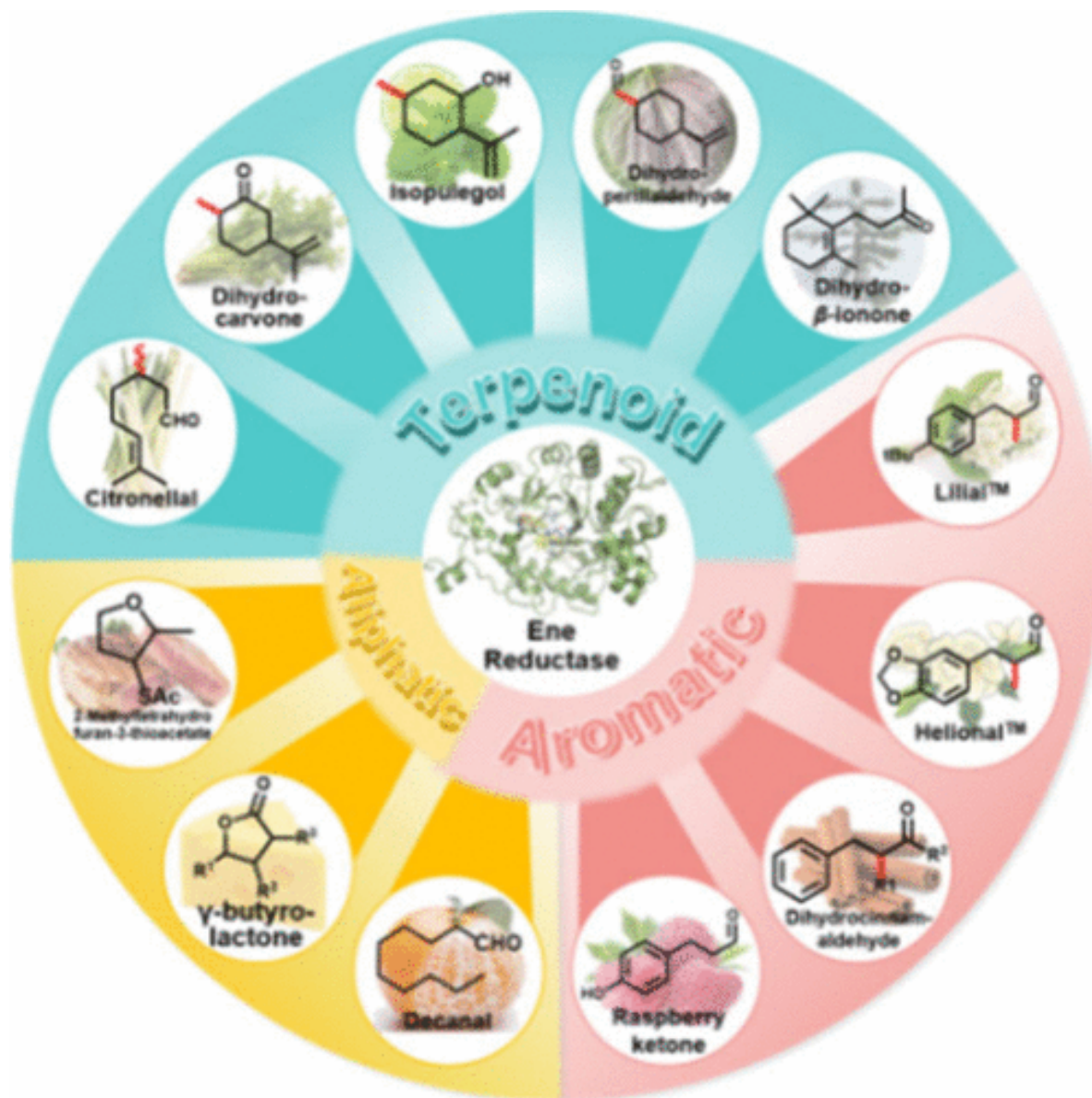
Classification of EREDs



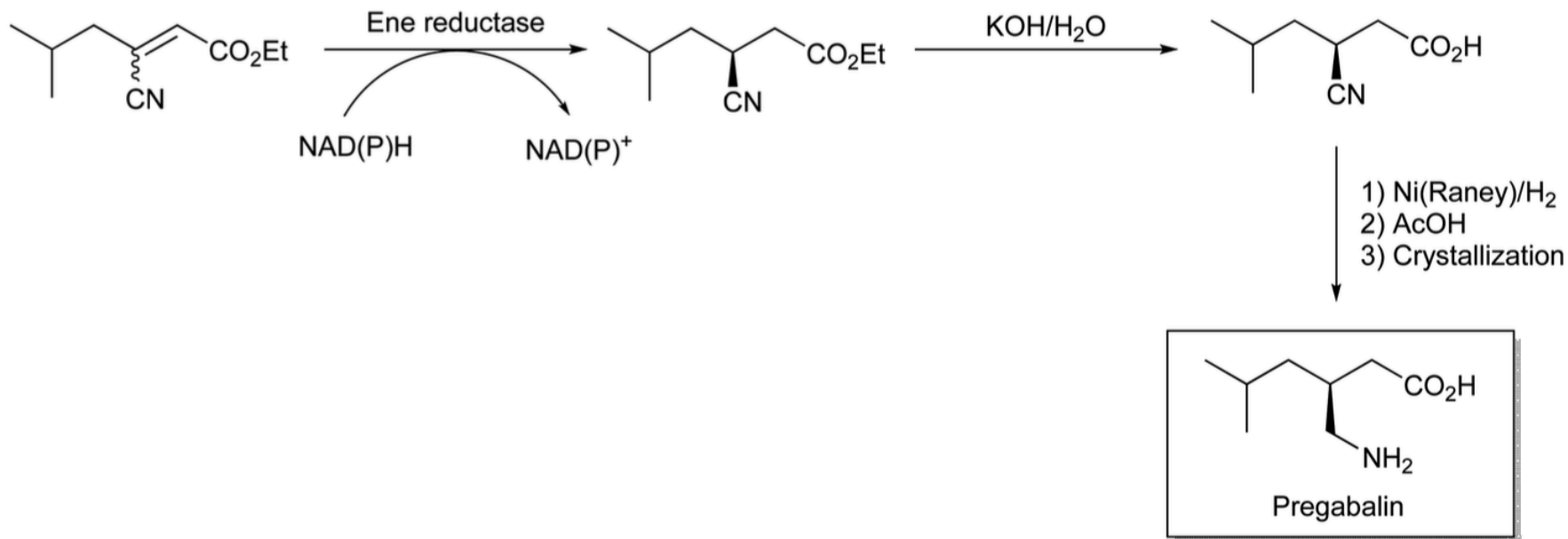
EREDs in Flavors and Frangrances



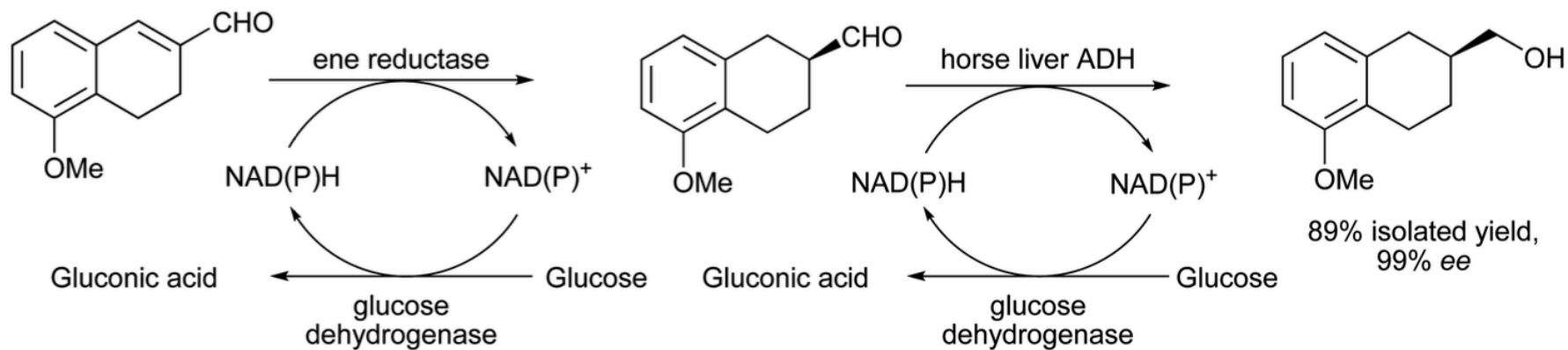
EREDs in Flavors and Fragrances



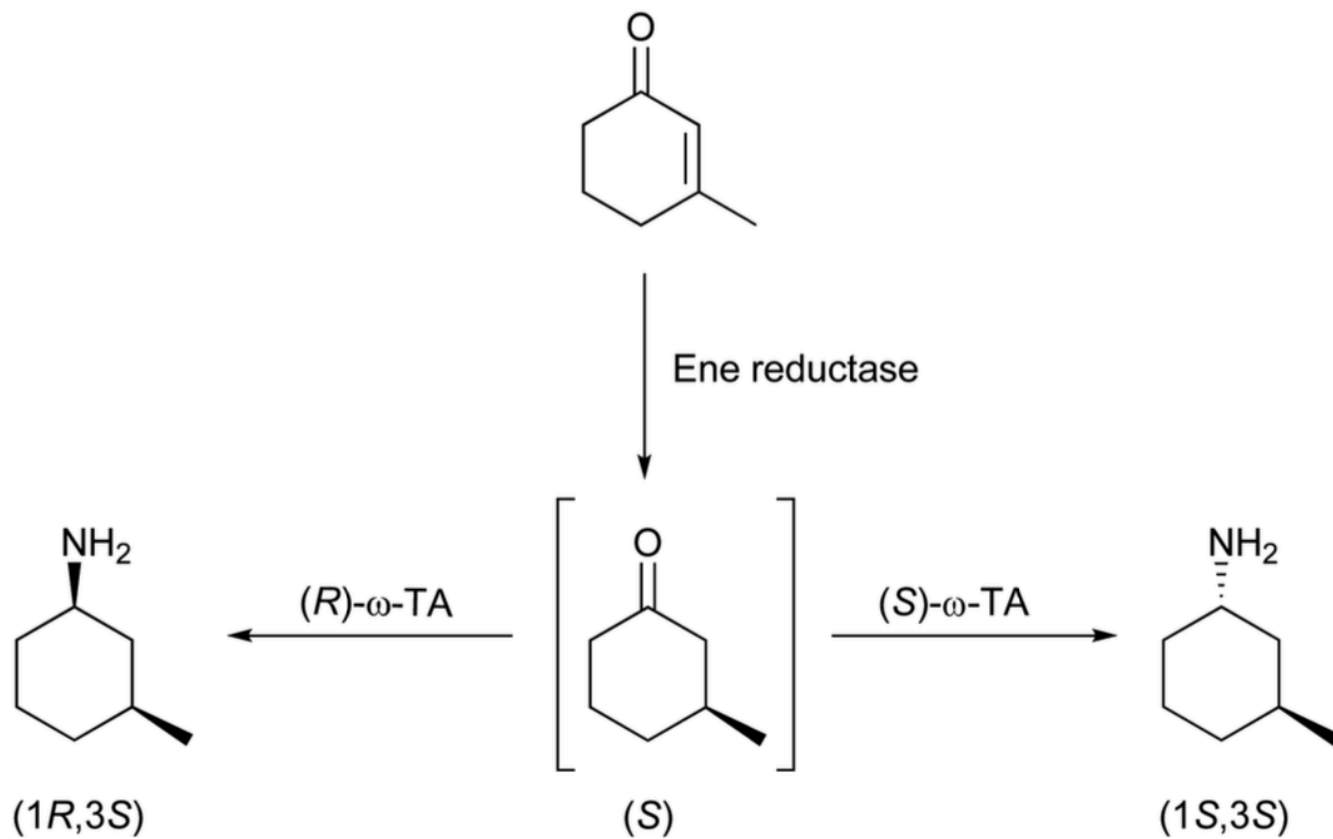
ERED Applications in Industry: Toward Pregabalin (Lyrica: Nerve Pain Medication)



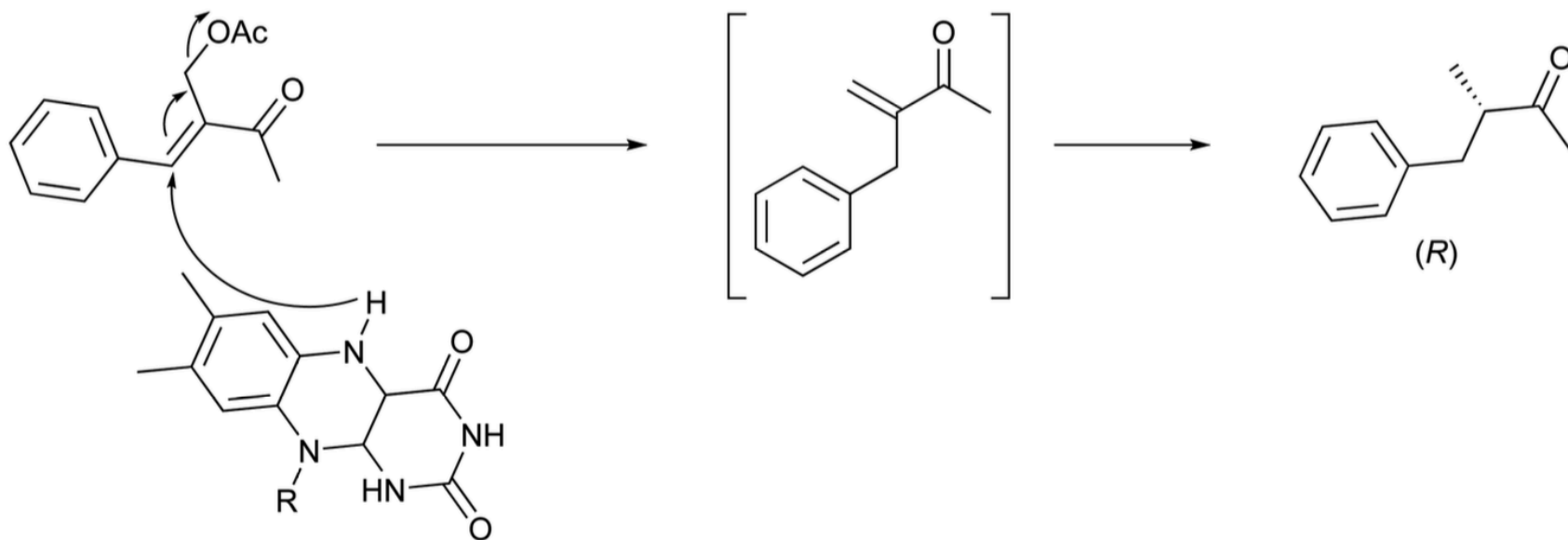
ERED Applications in Industry: Gluconic Acid to CNS Drugs



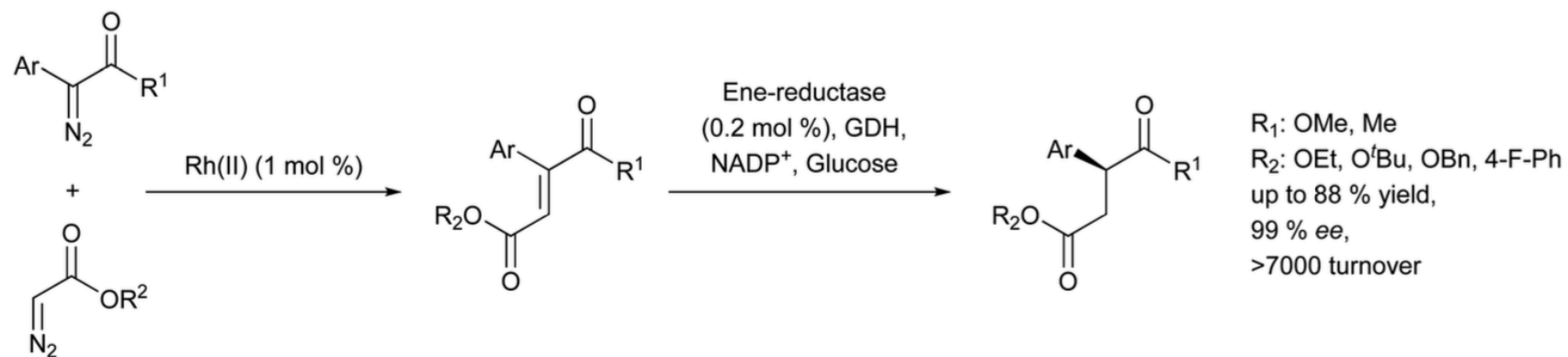
ERED Applications in Industry: Cascade Reactions with ATAs



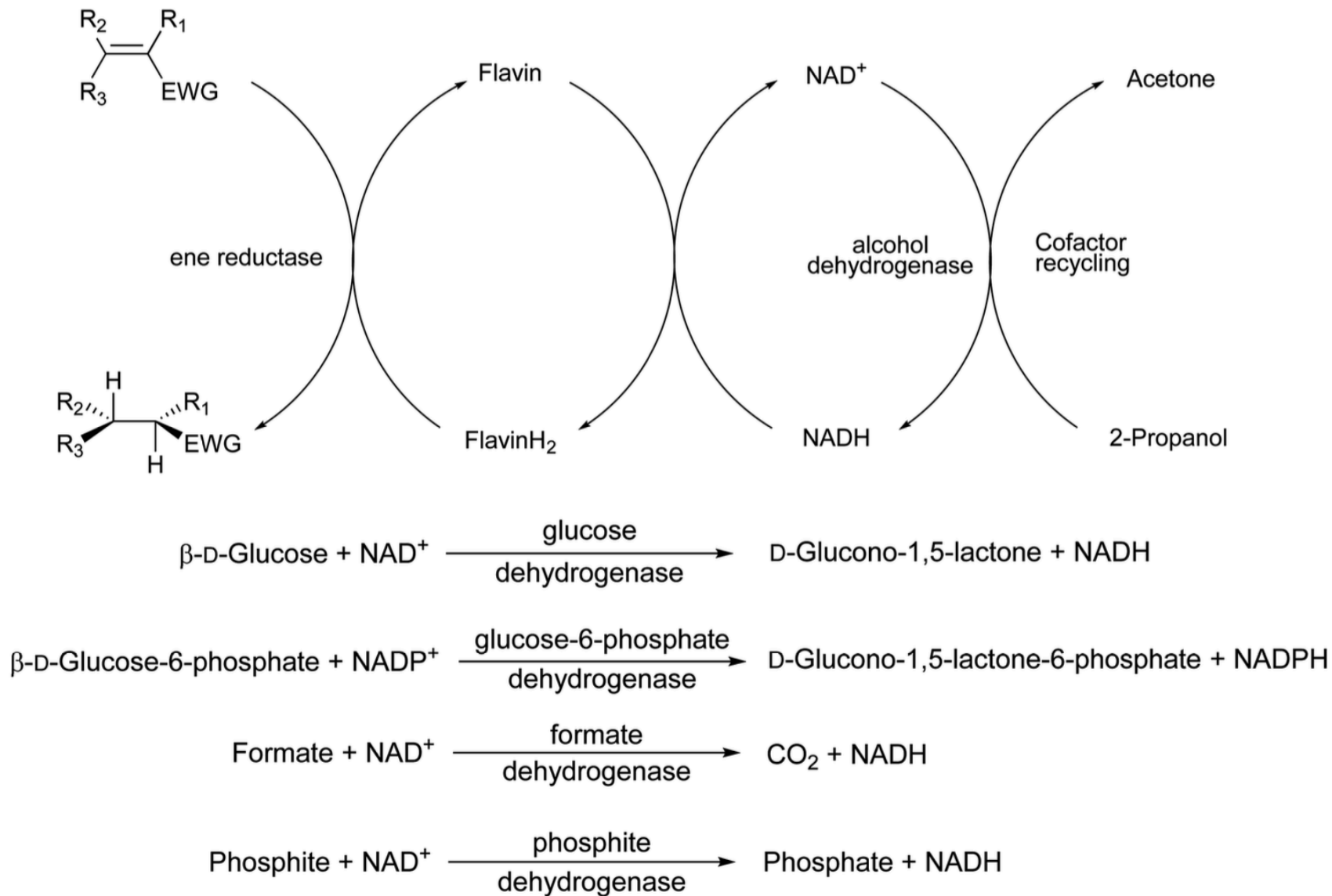
ERED Applications in Industry: Tandem S_N2' and Stereocenter Formation



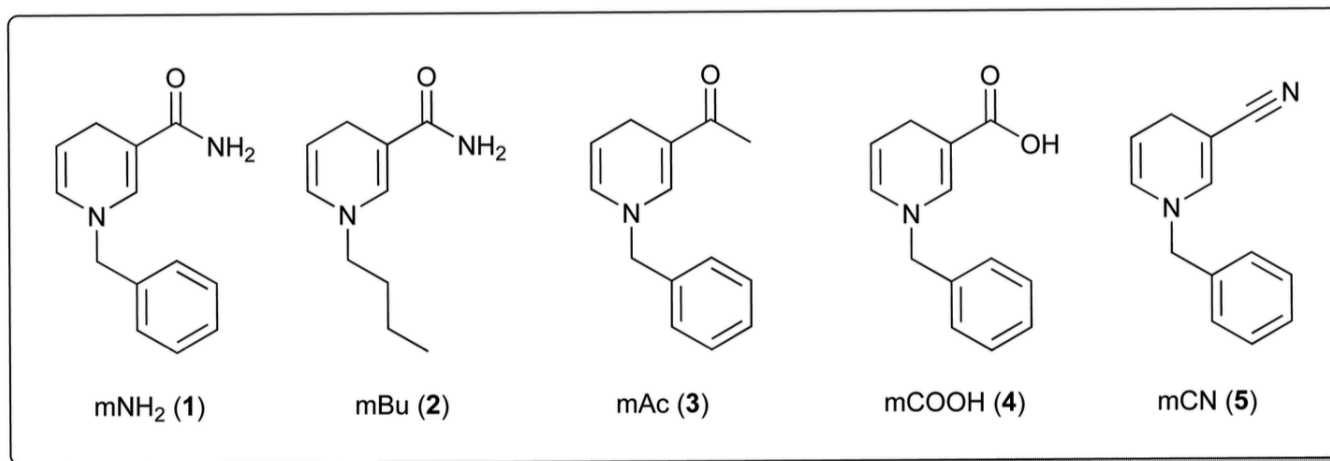
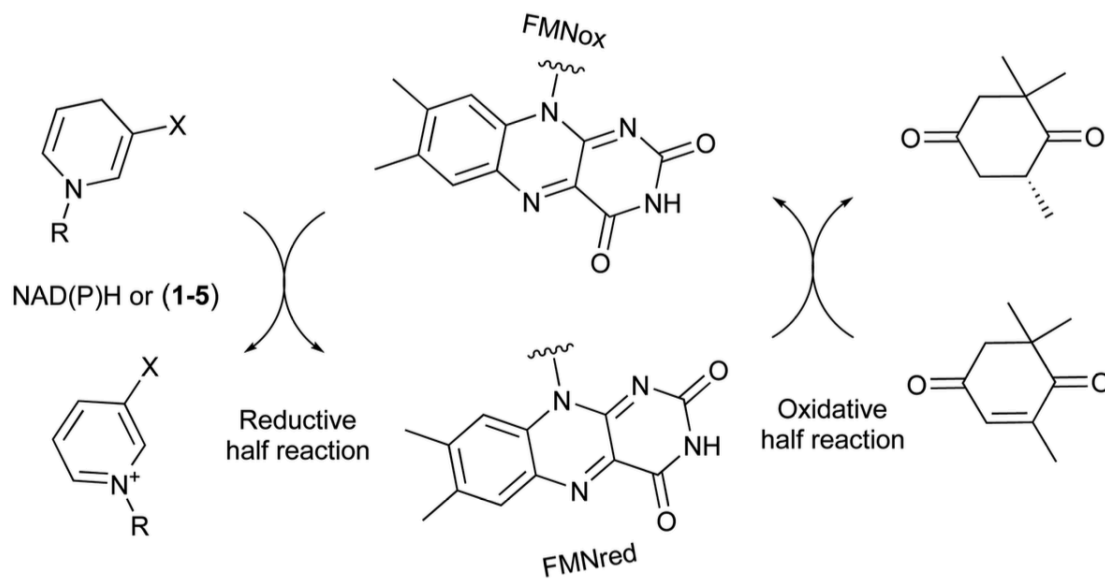
ERED Applications in Industry: Diazo Coupling and ERED Reduction



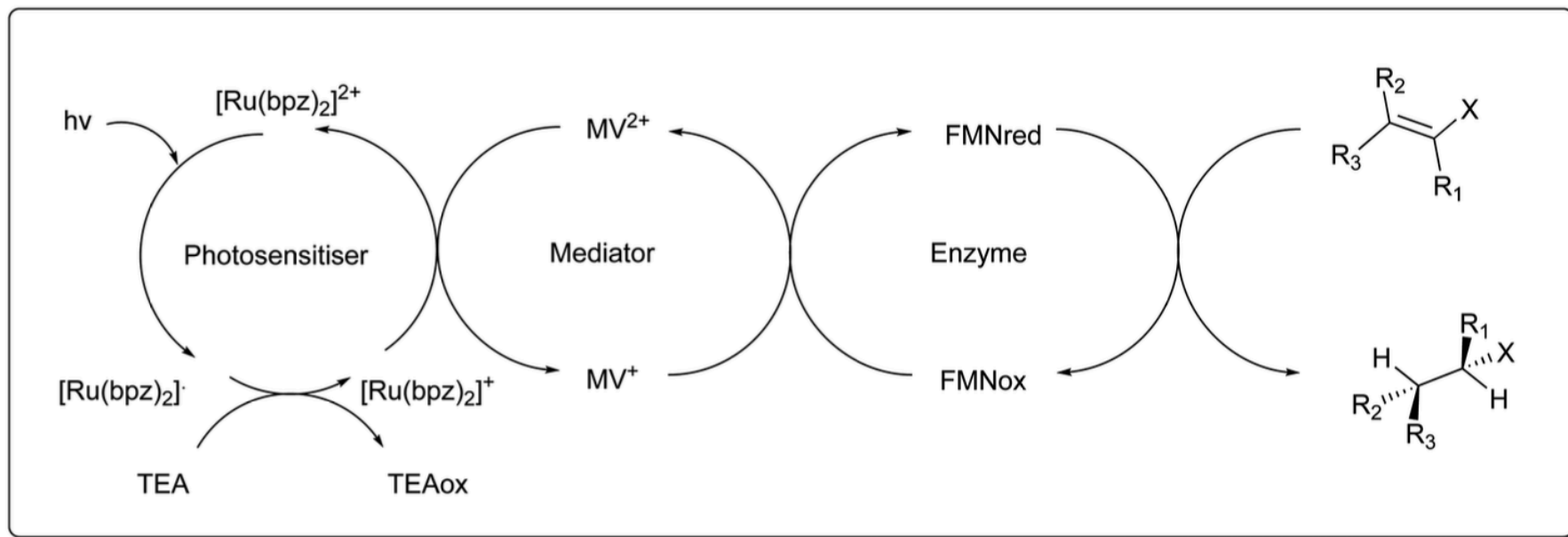
Turnover Systems for EREDs on Industrial Scale



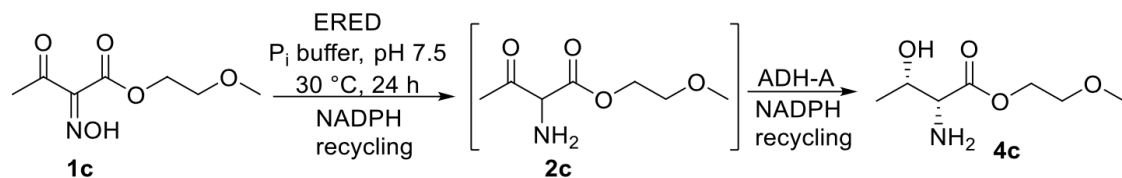
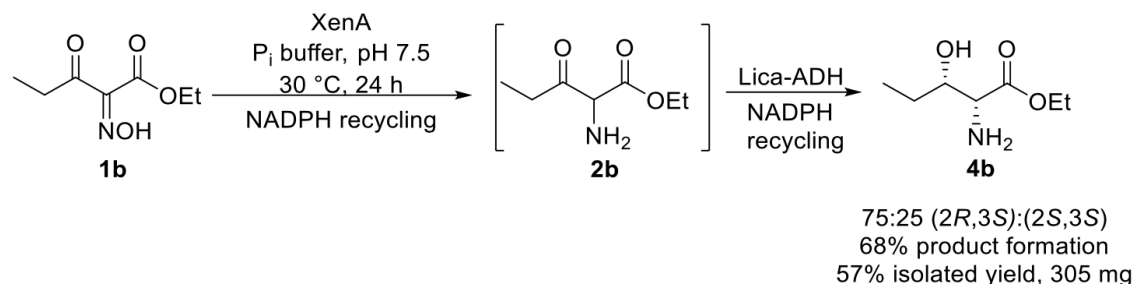
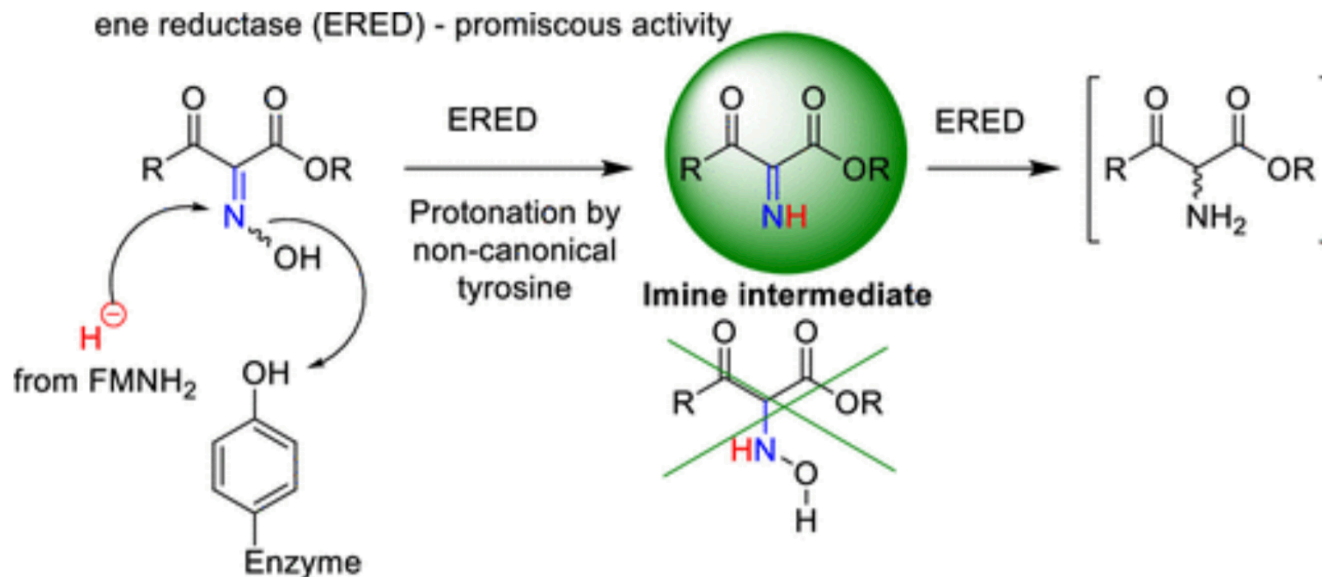
Common Turnover Systems to Avoid Nicotinamide: Biomimetics



Common Turnover Systems to Avoid Nicotinamide: Reduction Through Photoredox

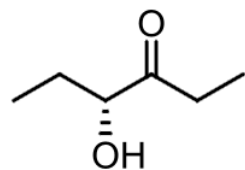
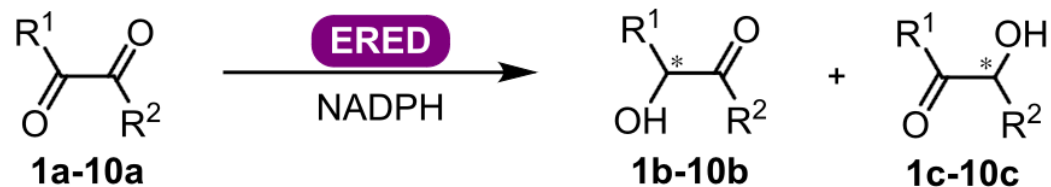


Emerging Applications of EREDs: Oxime Reduction



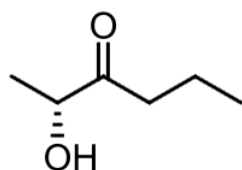
98% ee, 93% de for (2*R*,3*S*)
44% product formation (analytical scale, OPR3)
27% isolated yield, 154 mg (preparative scale, OYE3)

Emerging Applications of EREDs: ERED Becomes a KRED



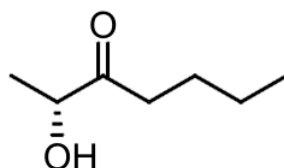
1b

GluER 8% ee: 92%



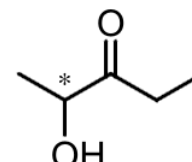
2b

GluER 19% ee: 83%



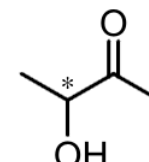
3b

GluER 17% ee: 67%



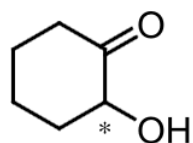
4b

GluER <1%



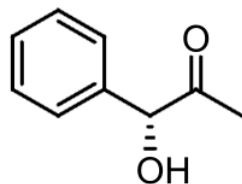
5b

GluER <1%



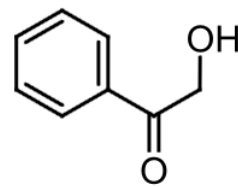
6b

OYE2 2%
TsOYE 2%



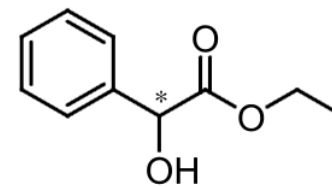
7b

GluER 37% ee: 79%
OYE3 91% ee: >99.9%
OYE2 79% ee: >99%
YqjM 2%
TsOYE 14% ee: 74%
NtDBR 6% ee: 21%



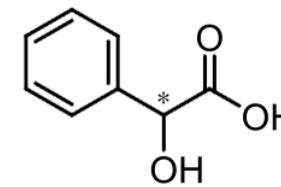
8c

GluER 93%
OYE3 86%
OYE2 62%
YqjM 55%
TsOYE 47%
NtDBR 30%



9b

0%

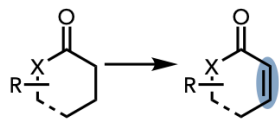


10b

0%

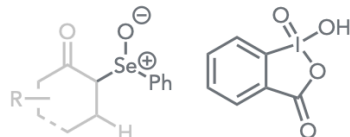
Emerging Applications of EREDs: Run it In Reverse for Desaturation

a Carbonyl Desaturation



wildly practiced
in organic synthesis

Classic Synthetic Methods



- high loadings of metal catalysts/strong oxidizing conditions
- stereoselective methods are rare

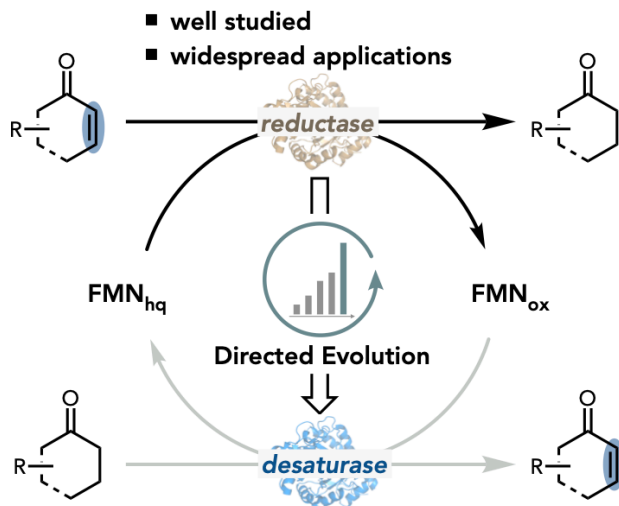


Enzymatic Methods *underexplored*



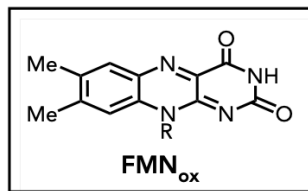
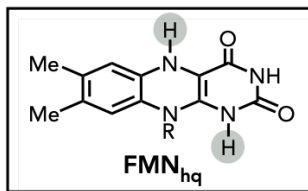
- complementary reactivity
- precise stereochemical control

b Repurposing 'Ene'-Reductases for Desaturation

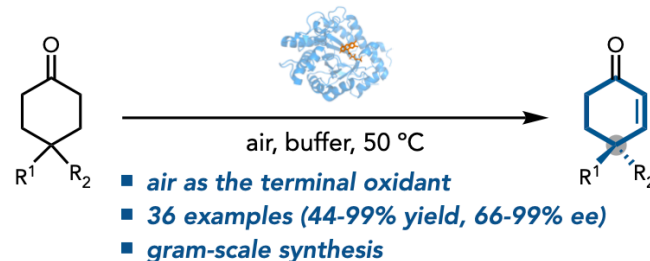


- well studied
- widespread applications

- Challenges:**
- thermodynamically disfavored process
 - enones are prone to disproportionation

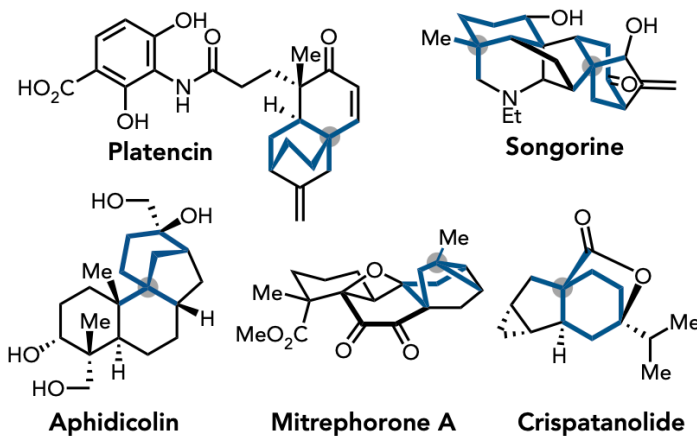


c This Work: Enzymatic Desymmetrizing Dehydrogenation



- air as the terminal oxidant
- 36 examples (44-99% yield, 66-99% ee)
- gram-scale synthesis

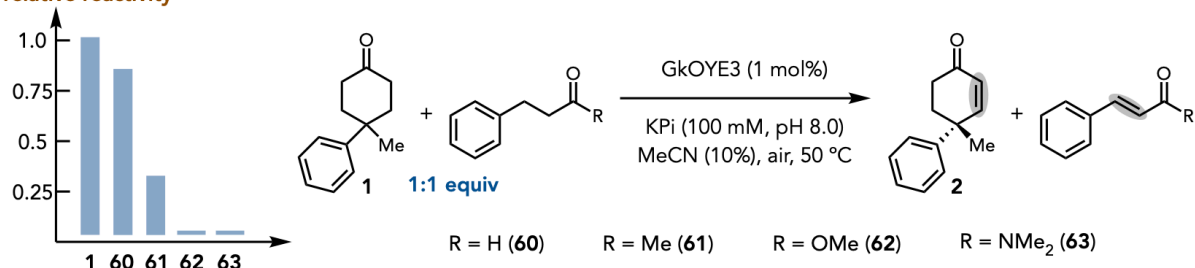
products as valuable chiral building blocks in total synthesis



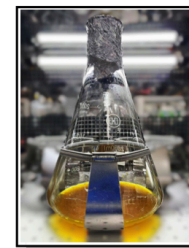
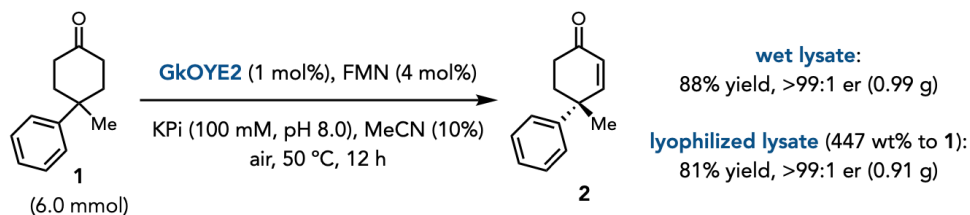
Emerging Applications of EREDs: Run it In Reverse for Desaturation

a Chemoselectivity between aldehyde, ketone, ester, and amide.

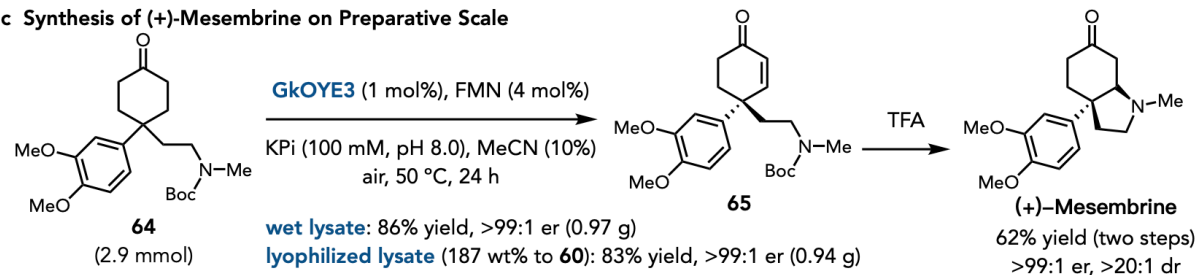
relative reactivity



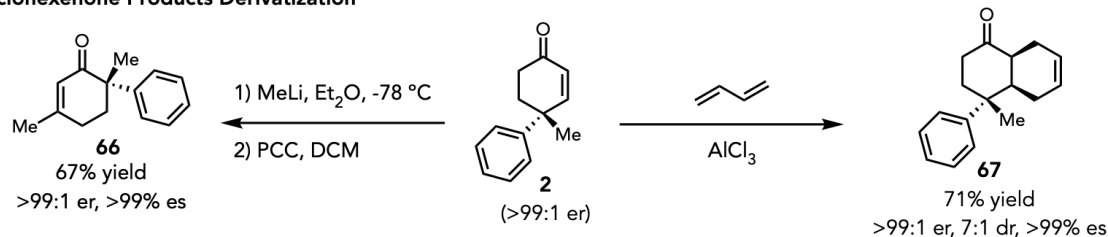
b Gram-Scale Preparation of 2 using Enzyme Wet Lysate



c Synthesis of (+)-Mesembrine on Preparative Scale



d Cyclohexenone Products Derivatization



Other Synthetic Applications of EREDs

